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hydrogen (H), 2 atoms of oxygen (O), and 3 atoms of sulfur (S). This is balanced because there is the same number of each type of atom on each side of the equation.

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Explanation: In answer C, each side contains 4 atoms of hydrogen (H), 2 atoms of ...

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Correct Answer: C. Single replacement
When hydrogen peroxide (H_2O_2) is

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added to potassium iodide (KI) solution, the hydrogen peroxide decomposes into water (H₂O) and oxygen (O₂). The chemical equation for this decomposition reaction is:

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by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Water.

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