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AP Chemistry Chapter 15 Answers - Zumdahl 15.65 $\text{HIn} \rightleftharpoons \text{In}^- + \text{H}^+ + \text{K} a = \frac{[\text{In}^-][\text{H}^+]}{[\text{HIn}]} = 1.0 \times 10^{-9}$ a. In a very acid solution, the HIn form dominates, so the solution will be yellow. b. The color change occurs when the concentration of the more dominant form is approximately ten times as great as the less dominant form of the indicator.

AP Chemistry Chapter 15 Answers - Zumdahl 15

A one-significant-figure number divided by a two-significant-figure number yields an answer with one significant figure (answer = 1). 24. From Figure 1.9 of the text, a change in temperature of 180°F is equal to a change in temperature of 100°C and 100 K.

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