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Macroscopic potential and kinetic energy then naturally add to the internal energy for the total energy. The first law of thermodynamics, which often is taken as synonymous with the energy equation, is shown as a natural consequence of the energy equation applied to a cyclic process.

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Gordon John Van Wylen (born February 6, 1920) is an American physicist and author of textbooks on thermodynamics. He was chair of the physics department at the University of Michigan from 1969 to 1972 and President of Hope College in Holland, Michigan, from 1972 to 1987. He was born in Grant, Michigan and turned 100 in February 2020.

Gordon Van Wylen - Wikipedia

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Sonntag, Borgnakke and van Wylen 2.5 An electric dip heater is put into a cup of water and heats it from 20°C to 80°C. Show the energy flow(s) and storage and explain what changes. Solution: Electric power is converted in the heater element (an electric resistor) so it becomes hot and gives energy by heat transfer to the water.

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