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ΔH (f) $-1 = -840$ kJ mol A. -190 B. -95 C. $+6$ D. $+95$ -1 7 2 5 7 -1 5 Which is a correct definition of lattice enthalpy? A. It is the enthalpy change that occurs when an electron is removed from 1 mol of gaseous atoms. B. It is the enthalpy change that occurs when 1 mol of a compound is formed from its elements.

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